

報 文

溶融技術による土壤等からのセシウム熱分離に関する プラント実証試験評価

釜田 陽介^{1*}、阿部 清一²、川本 克也³、由井 和子²、倉持 秀敏²、大迫 政浩²

¹株式会社クボタ 水・環境総合研究所（〒661-8567 兵庫県尼崎市浜1-1-1）

²国立研究開発法人 国立環境研究所 資源循環・廃棄物研究センター（〒305-8506 茨城県つくば市小野川16-2）

³岡山大学大学院 環境生命科学研究科（〒700-8530 岡山市北区津島中3-1-1）

Plant Demonstration Test on the Thermal Separation of Cesium by Melting Technique from Soil

Yosuke KAMATA^{1*}, Seiichi ABE², Katsuya KAWAMOTO³, Kazuko YUI²,
Hidetoshi KURAMOCHI², and Masahiro OSAKO²

¹Water & Environment R & D, KUBOTA Corporation (1-1-1 Hama, Amagasaki, Hyogo, 661-8567 Japan)

²Center for Material Cycles and Waste Management Research, National Institute for Environmental Studies
(16-2 Onogawa, Tsukuba, Ibaraki, 305-8506 Japan)

³Graduate School of Environmental and Life Science, Okayama University
(3-1-1 Tsushima-naka, Kita-ku, Okayama, 700-8530 Japan)

Summary

By the accident of Fukushima Daiichi nuclear power plant in March 2011, the east of Japan was widely contaminated by radioactive cesium (Cs). Since the accident, a variety of volume reduction technology for radioactively contaminated waste has been developed. Melting technique is considered to be one of useful high-temperature treatments carried at 1300-1400°C. Using the technique, Cs can be separated in high efficiency from different solids degreasing their volume significantly. In our previous laboratory test, a high Cs volatilization ratio had been obtained for various solid wastes employed. Therefore, in this study, we conducted a demonstration test using a melting test plant (3 t/day) and investigated the Cs separation performance for soil, biomass incineration ash, and sewage sludge incineration ash. As a result, regardless of the kind of solid sample, Cs volatilization ratio increased by the addition of CaCl₂. Hence, CaCl₂ was considered to have a promoting effect on the Cs volatilization even in the plant scale test. The Cs volatilization ratio in the presence of combustible vegetation was higher than that without the combustibles. This shows the coexistence-effect of combustibles on the Cs volatilization. Increasing the amount of additive reagent and vegetation achieved 99.9% of the Cs volatilization ratio. Furthermore, using the two series of Bag Filter (BF) units, Cs was concentrated into the first (No.1) BF ash, and the amount of the Cs-concentrated ash was reduced by 1/3 to 1/4 compared with the case of using one BF equipment. The No.1 BF ash was primarily composed of alkali metal chloride, and Cs in the No.1 BF ash was highly water-soluble. This indicates that the secondary treatment of Cs by wet process (water elution - solid-liquid separation - Cs adsorption) is possible. The second (No.2) BF ash was composed of neutralized products of acid components (HCl and SO_x) in flue gas as follows; CaClOH, CaCl₂·4H₂O, CaCO₃ and Ca(OH)₂. Therefore, the No.2 BF ash is considered to be reusable as the melting addition reagent.

Key Words: Cesium, Soil, Melting, Thermal separation, Chlorination volatilization

和文要約

1300～1400 °Cでの高温熱処理により、各種固形物からCsを塩化揮発等により高度に分離濃縮、減容化できる溶融技術について、プラント試験レベルでのCs分離実証を目的とし、溶融テストプラントを用いて土壤、バイオマス焼却灰、下水汚泥焼却灰等、様々な固形物を処理対象とした試験を行った。その結果、処理対象に関わらずCaCl₂の添加によりCs揮散率は増加し、プラント実証レベルでもCaCl₂はCsの塩化揮発促進効果を有することが実証された。また草木類を含む条件の方がCs揮散率は高く、可燃物の共存による揮散促進効果も確認された。添加薬剤と草木類の量を増加させることにより、最大99.9%のCs揮散率が得られた。また、二段BFの適用によりCsはNo.1 BF灰に濃縮され、一段BFの場合に対して1/3～1/4に減量した。No.1 BF灰は、アルカリ金属塩化物が主体であり、水溶性が高かったため、湿式処理(水溶解-固液分離-Cs吸着)によるCsの二次濃縮、保管が可能であると考えられた。No.2 BF灰は、純粋な酸性ガス中和物であり、CaClOH, CaCl₂·4H₂O, CaCO₃, Ca(OH)₂で構成されたため、溶融薬剤として循環再利用が可能であると考えられた。